



*Norwegian
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EMEP model: Chemical modules

David Simpson

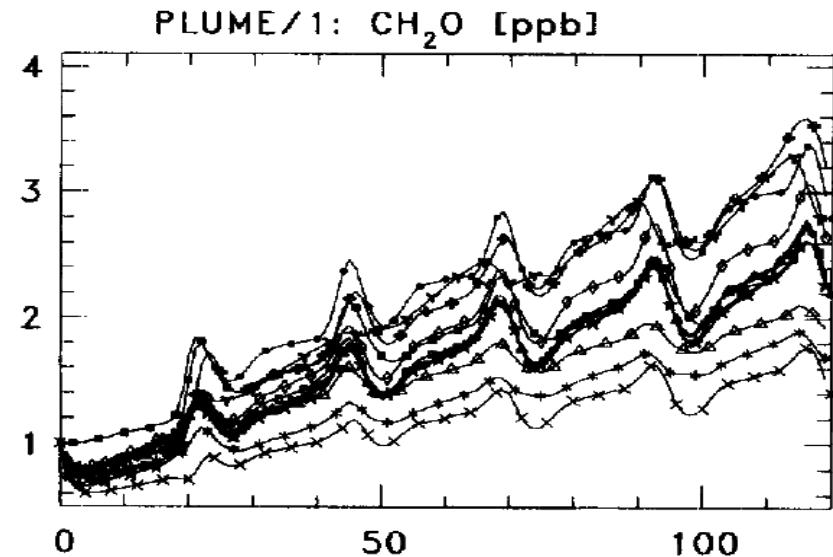
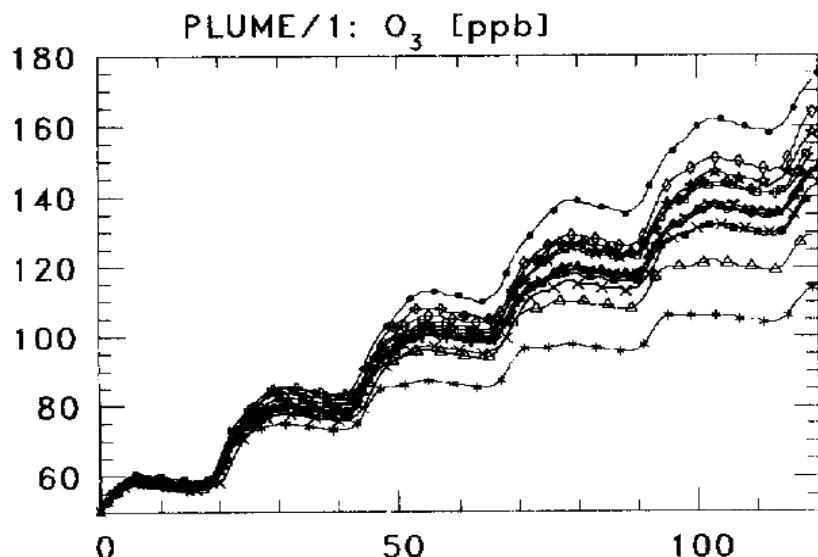


Outline:

- Gas-phase chemistry
- SOA module
- Code structure
- Miscellaneous (mass balance, GenChem, other schemes)

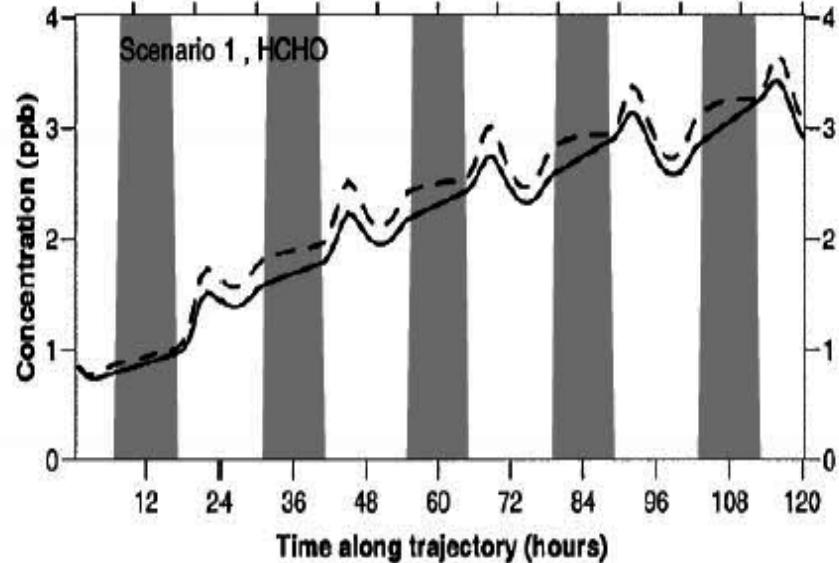
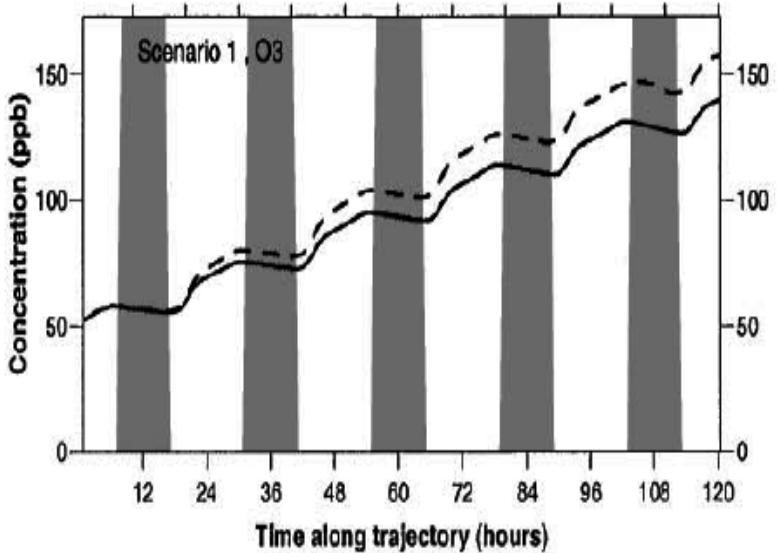


Early comparisons.....



Kuhn et al., Atm.Env., 1998, with EMEP, IVL, 4 x RADM, 3 x CB4, ...

Follow-up: EMEP vs most complex (IVL): comparisons.....



Andersson-Sköld & Simpson, Atm.Env., 1999, same setup as Kuhn et al. (+other cases)

- EMEP model compared well :-)



Gas-phase chemistry:

- EmChem09
 - 72 species
 - 137 reactions
 - 10 VOC (1 BVOC = isoprene)
- Update of previous EMEP chemical schemes, which date back to Eliassen, Hov, et al 1982
- 'Surrogate species' method, with relatively more RO₂ chemistry than e.g. CB schemes



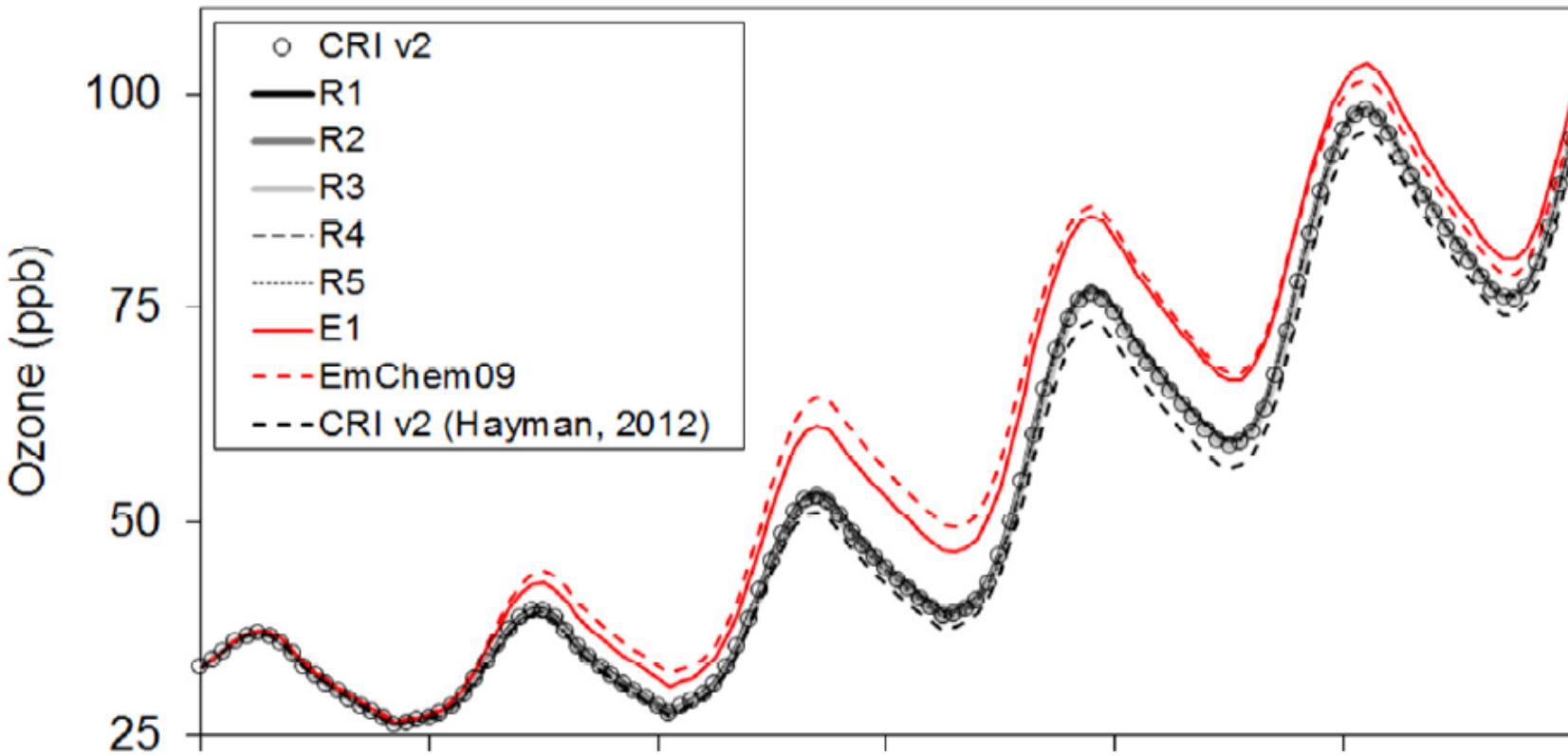
EmChem09:

- **Updates 2008-2009:**

- compared against UK MCM, CRI, OSRM and US CB4, CB5 mechanisms
- R. Bergström, G. Hayman, M. Jenkin, C. Richter and D. Simpson
 - EMEP scheme compared well to CRI, although slightly “hot”.



Comparison with CRI model.

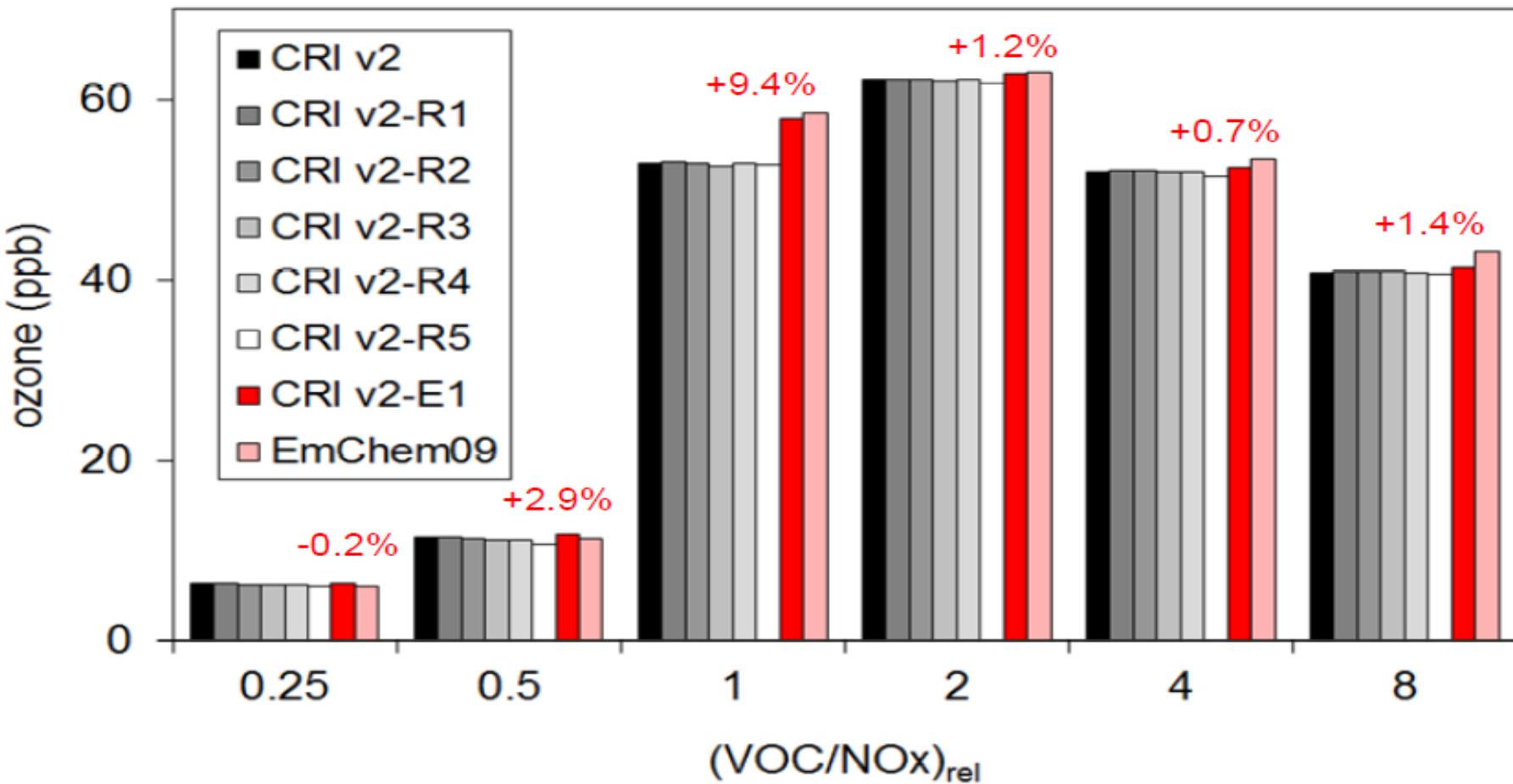


Ozone changes (c/o Mike Jenkin), simple plume

- CRI v2 ~ 1000 reactions, 112 AVOC



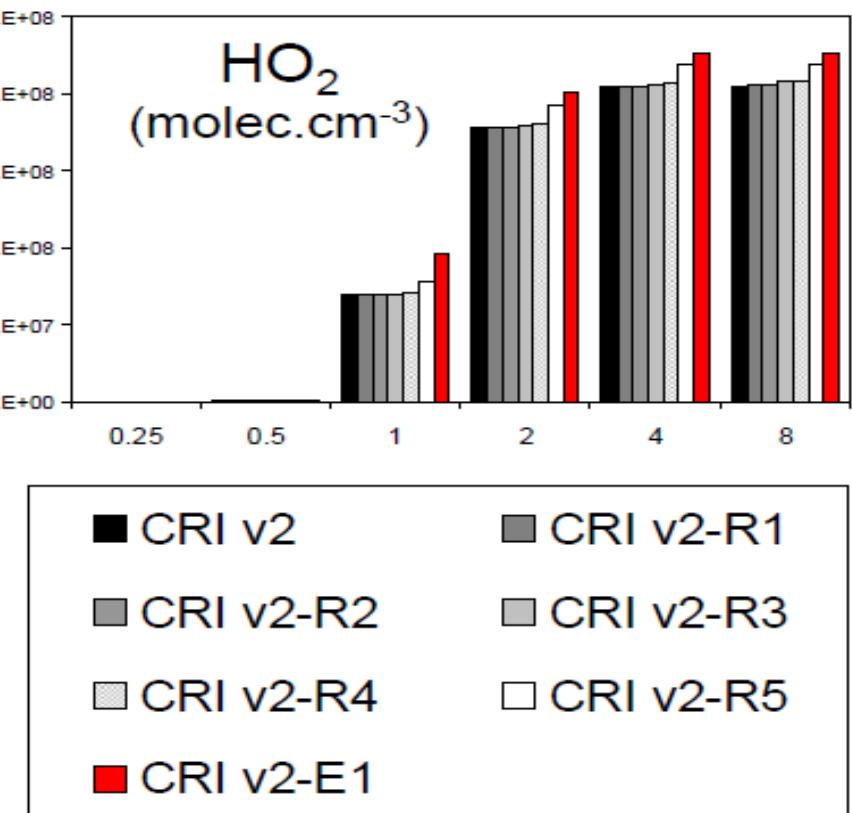
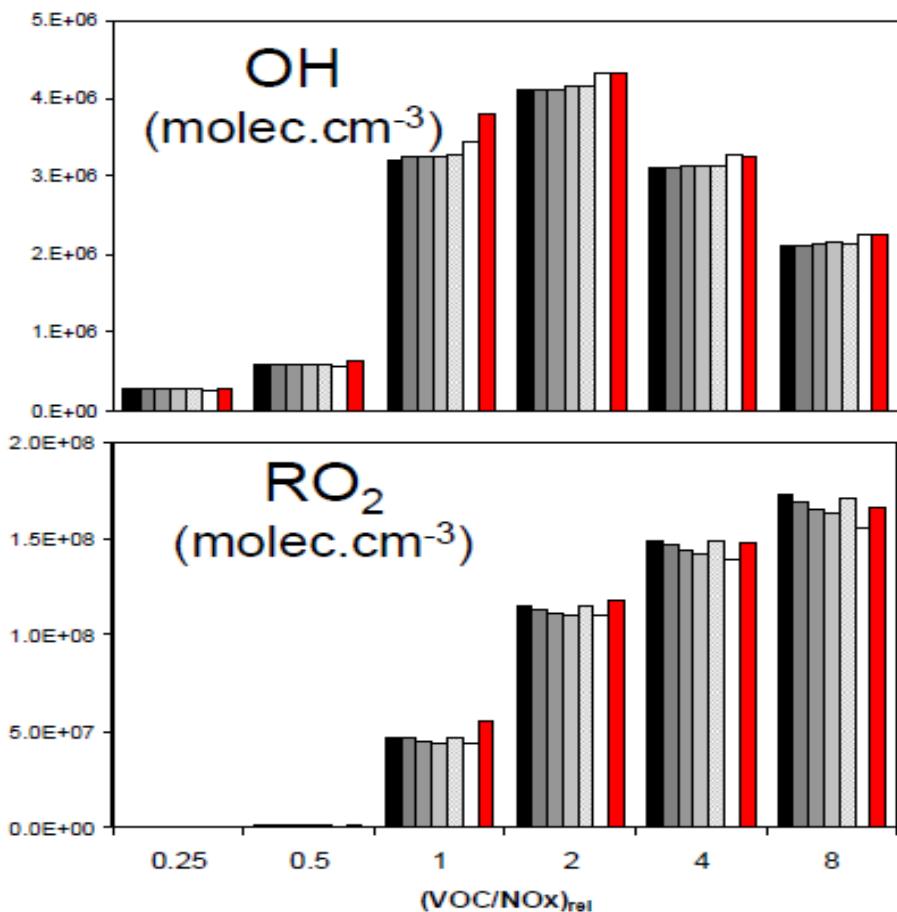
Comparison with CRI model.



Ozone at different VOC/NOx ratios (5-day mean)
- (c/o Mike Jenkin)



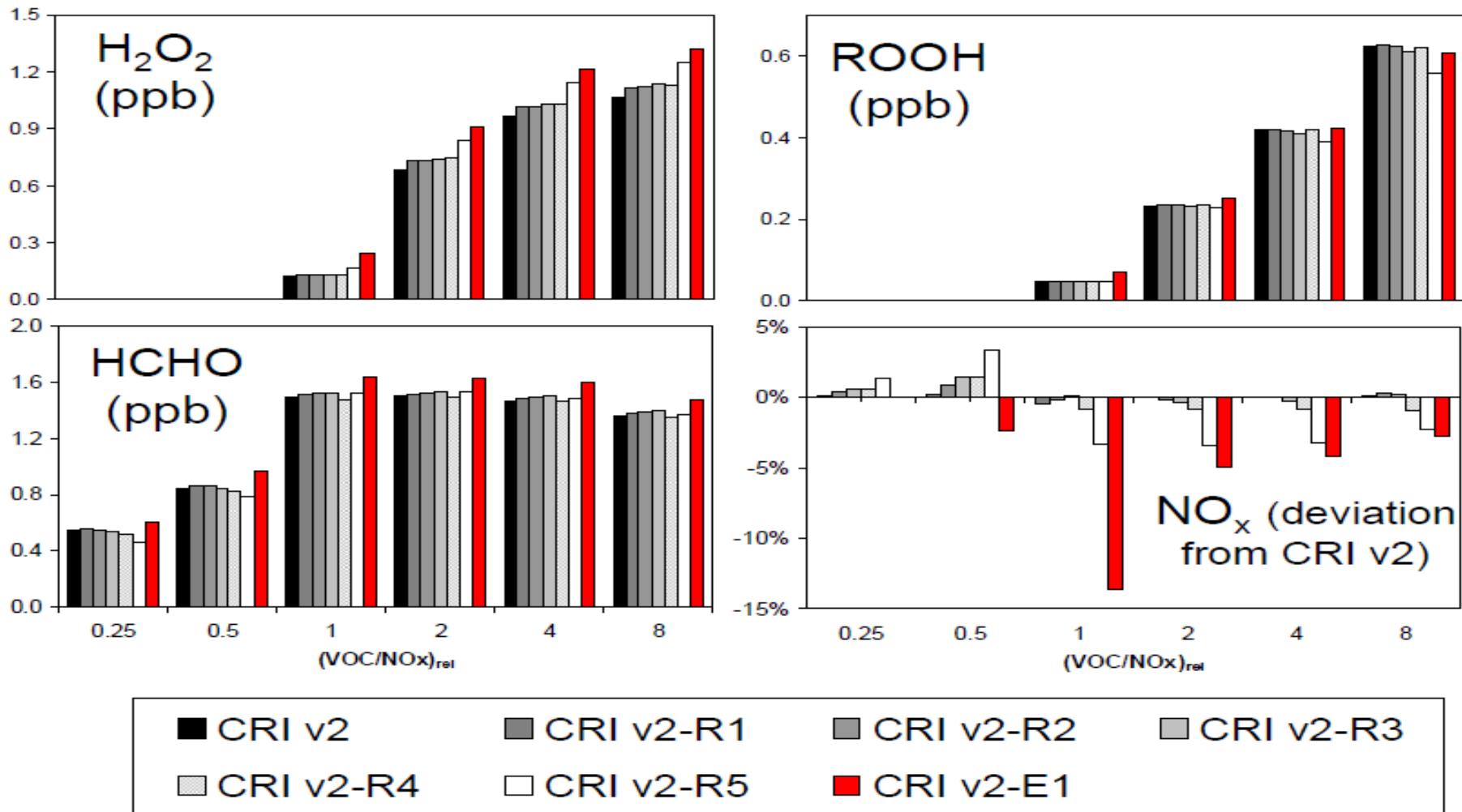
Comparison with CRI model.



Radicals (c/o Mike Jenkin)

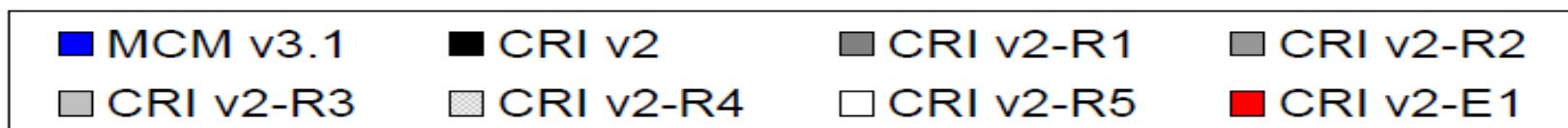
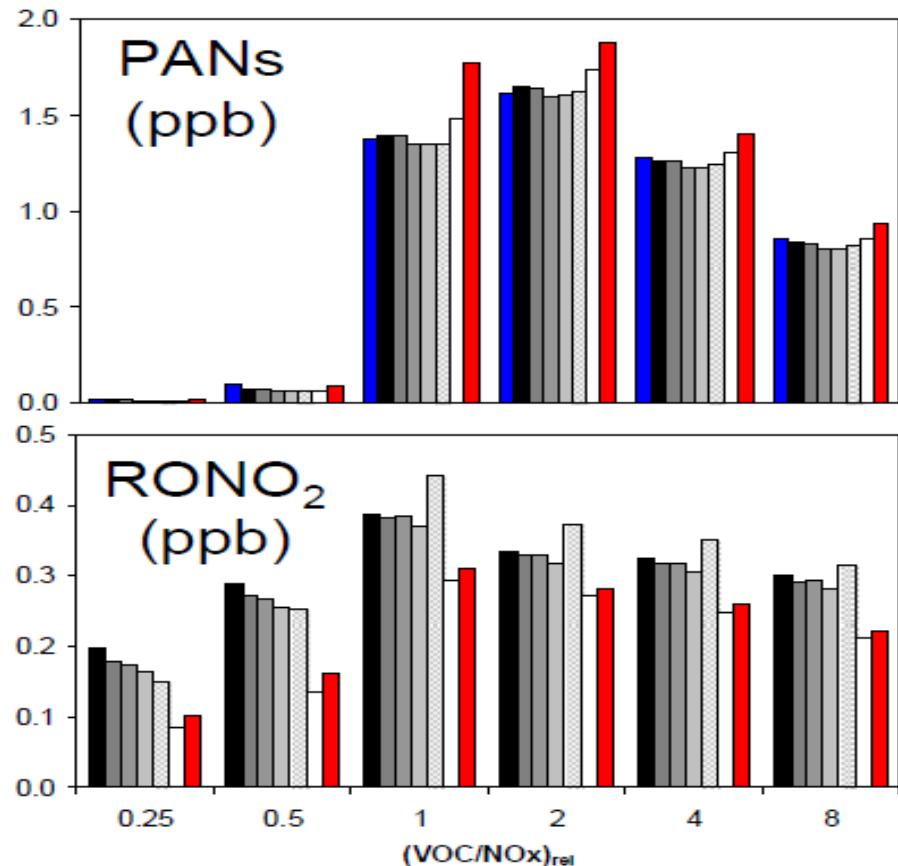
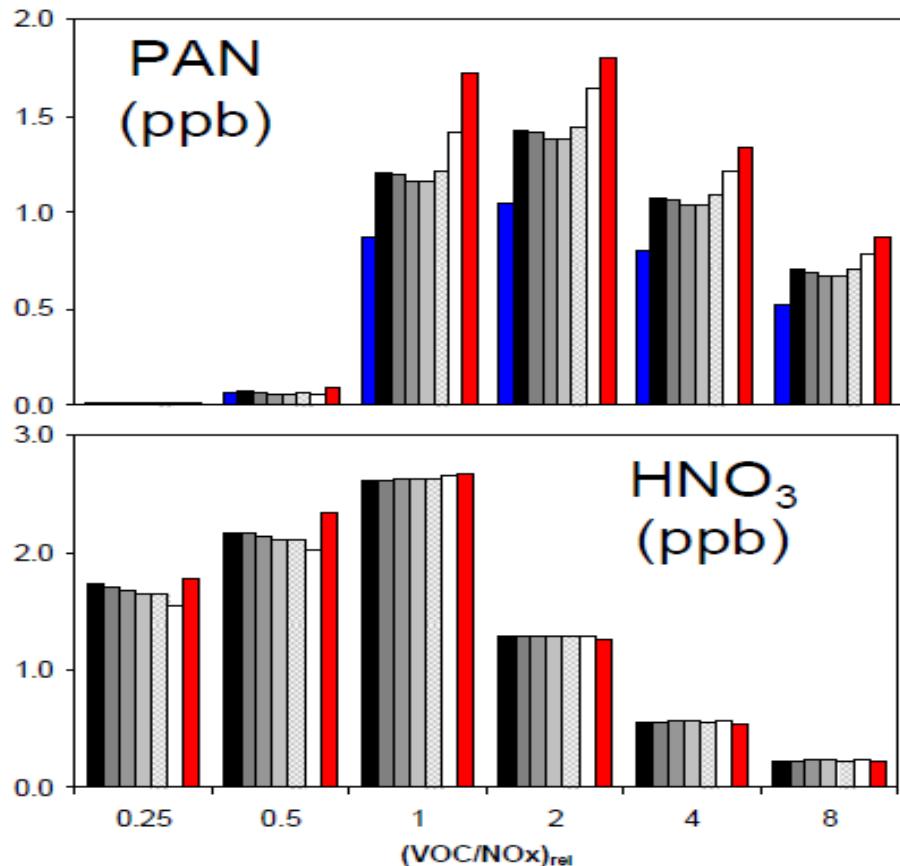


Comparison with CRI model.





Cont.





How do we handle chemistry in the model?

- Solver_ml + CM_ files:



•CM_ files:

```
ChemFunctions_ml.f90
Chem_ml.f90
CM_BoundaryConditions.inc
CM_BoundaryConditions.inc.txt
CM_ChemGroups_ml.f90
CM_chempackages.txt
CM_ChemRates_ml.f90
CM_ChemSpecs_ml.f90
CM_DryDep.inc
CM_EmisBioNat.inc
CM_EmisFiles.inc
CM_emislist.csv
CM_EmissSpecs.inc
CM_Reactions1.inc
CM_Reactions2.inc
CM_WetDep.inc
CoDep_ml.f90
```



• CM_ChemSpecs_ml

•

• Species -

- Number
- Indices

-

```
!>U

      module ChemSpecs_adv_ml
!-----

      ...
      integer, public, parameter :: NSPEC_ADV = 126

      integer, public, parameter :: IXADV_03      = 1    &
      , IXADV_NO       = 2    &

      ...
!-----
      end module ChemSpecs_adv_ml
!>

      module ChemSpecs_shl_ml
!-----


      ...
      IXSHL_OD      = 1    &
      , IXSHL_OP     = 2    &

      ...
!-----
      module ChemSpecs_tot_ml
!-----


      integer, public, parameter :: NSPEC_TOT = 142

      integer, public, parameter :: OD      = 1    &
      , OP       = 2    &
```



• CM_ChemSpecs_ml

• + characteristics

```
...  
contains  
subroutine define_chemicals()
```

```
!  
species(OD ) = Chemical("OD ", 16.0, 0, 0, 0, 0, 0.0, 0.0, 0.0, 0.0)  
species(OP ) = Chemical("OP ", 16.0, 0, 0, 0, 0, 0.0, 0.0, 0.0, 0.0)  
species(OH ) = Chemical("OH ", 17.0, 0, 0, 0, 0, 0.0, 0.0, 0.0, 0.0)  
species(HO2 ) = Chemical("HO2 ", 33.0, 0, 0, 0, 0, 0.0, 0.0, 0.0, 0.0)  
  
species(O3 ) = Chemical("O3 ", 48.0, 0, 0, 0, 0, 0.0, 0.0, 0.0, 0.0)  
species(NO ) = Chemical("NO ", 30.0, 0, 0, 1, 0, 0.0, 0.0, 0.0, 0.0)  
species(NO2 ) = Chemical("NO2 ", 46.0, 0, 0, 1, 0, 0.0, 0.0, 0.0, 0.0)  
species(PAN) = Chemical("PAN", 121.0, 0, 2, 1, 0, 0.0, 0.0, 0.0, 0.0)  
species(NO3 ) = Chemical("NO3 ", 62.0, 0, 0, 1, 0, 0.0, 0.0, 0.0, 0.0)
```



•CM_Reactions1.inc

! -> 03

```
-  
P = &  
    rct(1,k) * xnew(OP ) &  
    + 0.15*rct(37,k) * xnew(CH3C002 ) * xnew(H02 ) &  
    + 0.29*rct(60,k) * xnew(MAC03 ) * xnew(H02 )  
  
L = &  
    rct(5,k)* xnew(N0 ) &  
    + rct(6,k)* xnew(N02 ) &  
    + rct(7,k)* xnew(OH ) &  
    + rct(8,k)* xnew(H02 ) &  
    + AQRCK(ICLRC2,K)* xnew(S02 ) &  
    + rct(43,k)* xnew(C2H4 ) &  
    + rct(44,k)* xnew(C3H6 ) &  
    + rct(50,k)* xnew(C5H8 ) &  
    + rct(53,k)* xnew(MACR ) &  
    + rct(58,k)* xnew(MVK ) &  
    + rcphot(IDA03,K) &  
    + rcphot>IDB03,K  
  
xnew(03)= ( xold(03) + dt2 * P) /(1.0 + dt2*L )
```



•CM_ChemGroups_ml

```
!> -----
module ChemGroups_ml
!-----

use ChemSpecs_tot_ml ! => species indices
use OwnDataTypes_ml   ! => typ_sp
implicit none
private
! Assignment of groups from GenIn.species:
public :: Init_ChemGroups

! ----- Gas/particle species -----
..

integer, public, parameter :: INDEX_WDEP_OXN_GROUP = 2
integer, public, target, save, dimension(4) :: &
    WDEP_OXN_GROUP      = (/ HN03,HONO,N03_F,N03_C /)

integer, public, parameter :: INDEX_WDEP_PPM10_GROUP = 3
integer, public, target, save, dimension(11) :: &
    WDEP_PPM10_GROUP     = (/ POM_F_WOOD,POM_F_FFUEL,
UEL_NEW,EC_F_FFUEL_AGE,EC_C_FFUEL,REMPPM25,REMPPM_C /)
```



OA, POA, SOA ;-)

OA: Subject=Horrendous!

Some issues:

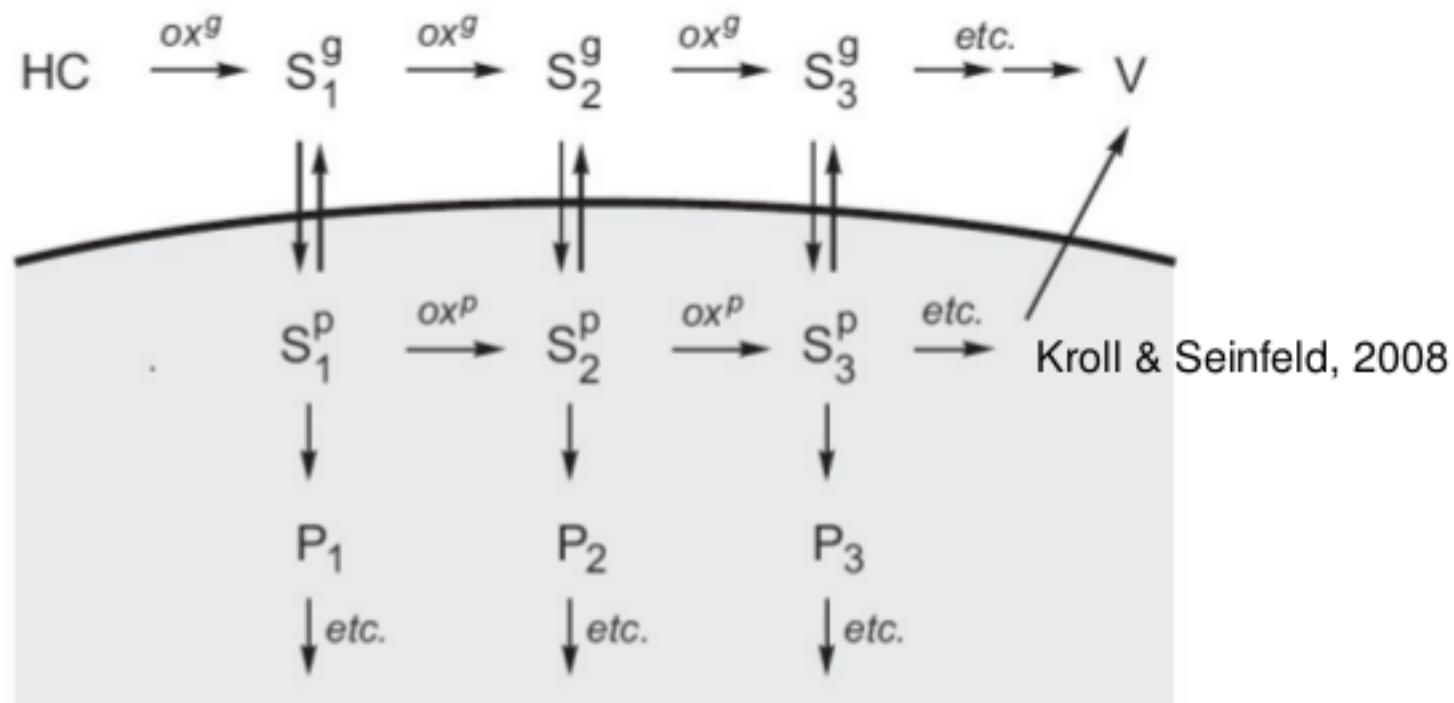
- Which precursors? (Isoprene, terpenes, aromatics ..)
- 1000s of compounds, mainly unknown.
- Gas-phase? Aqueous-phase? Particle-phase?
- Thermodynamics/Volatility
- Aging
- Artifact-rich observations!

See e.g. Donahue et al., Atmos. Environ., 2009, Hallquist et al., ACP, 2009



OA cont,

- Need to deal with equilibria, e.g.





OA cont., “VBS”:

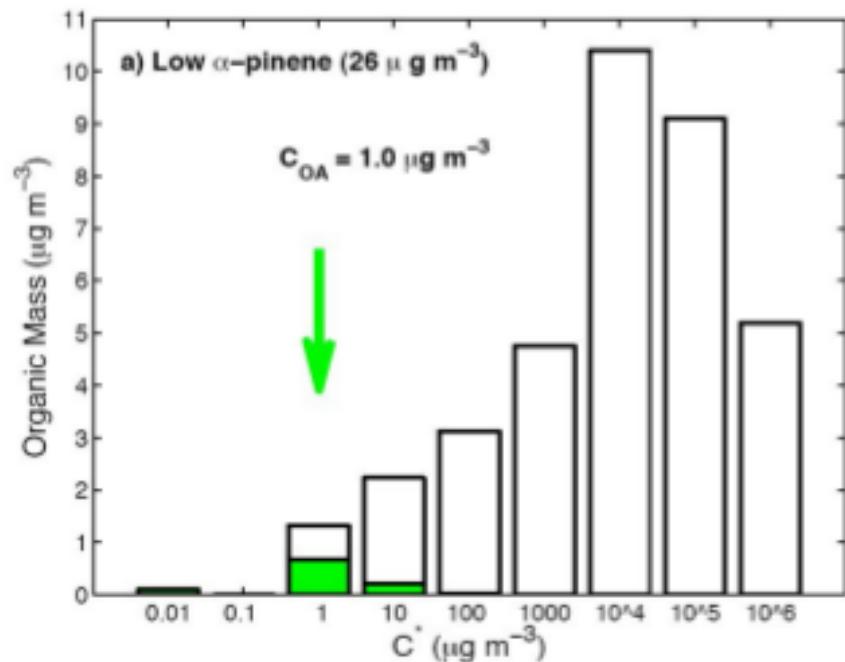
- Volatility Basis Set (Donahue and colleagues)

- Gas-Particle partitioning:

$$\frac{A_i}{G_i} = \frac{C_{OA}}{C_i^*}$$

where

- C_i^* is saturation concentration,
 $= f(\text{Vapour pressure})$





OA in model code:

- Code tracks species, not phase, for advection
- Gas/aerosol fractions calculated using partitioning theory in My_SOA_ml
- Affects deposition, aging reactions etc.
-



OA in model code:

- Origin (GenChem) of reactions looks like:

```
1.2e-11*exp(444.0*TINV)      [OH] + apinene = TERPPeroxy  
KRO2NO TERPPeroxy + NO = NO2 + 0.08 BSOC_ug1 + 0.672 NON_C_BSOA_ug1  
                         + 0.813333 BSOC_ug10 + 6.832 NON_C_BSOA_ug10  
                         + 1.34 BSOC_ugle2 + 11.256 NON_C_BSOA_ugle2  
                         + 3.333333 BSOC_ugle3 + 28. NON_C_BSOA_ugle3
```

- ie produces range of products of different volatility



OA in CM_Reactions2:

- Gas-phase chemistry

```
! -> BSOC_UG1

P =  &
    0.002833*rct(40,k)*xnew(ISR02)*xnew(N0)  &
    + 0.0255*rct(47,k)*xnew(ISR02)*xnew(H02)  &
    + 0.08*rct(40,k) * xnew(TERPPER0XY ) * xnew(N0 )  &
    + 0.715333*rct(75,k) * xnew(TERPPER0XY ) * xnew(H02 )  &
    + rct(85,k)*xnew(OH) * xnew(BSOC_UG10 )
]

L =  &
    rct(84,k)*xnew(OH)

xnew(BSOC_UG1)= ( xold(BSOC_UG1) + dt2 * P) /(1.0 + dt2*L )
```



OA final words: •SOA modules can give many very different answers.



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The basic rule:



Garbage in ⇒ ⇒ Garbage out:



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The basic rule:



Garbage in ⇒

⇒ Garbage out:

SOA twist:

Garbage in the middle!





OA: garbage avoidance...



Strategies:

- Check basics - does the model work for anything?
 - Check other pollutants - SO₂, SO₄, NO_x, NO_y,
- Check emissions!
- Check PCM tracers - EC, levoglucosan, C₁₄
- Check measurements - what do they mean?!
- Be humble.....



Miscellaneous 1: Mass balance

- Please check : fracmass in RunLog

```
+++++
Mass balance 1 Sulphur
]+++++
          sumint      summas      fluxout      fluxin      fracmass
Sulphur   3.4648E+07  4.2275E+07  5.2910E+09  3.6425E+09  1.0117E+00
          ifam       totddep     totwdep      totem
          1         1.795E+09   4.573E+09   7.888E+09
+++++
Mass balance 2 Nitrogen
]+++++
          sumint      summas      fluxout      fluxin      fracmass
Nitrogen  7.3405E+07  1.3389E+08  1.1912E+10  8.8174E+09  1.0078E+00
          ifam       totddep     totwdep      totem
          2         5.729E+09   1.245E+10   2.110E+10
+++++
Mass balance 3 Carbon
]+++++
          sumint      summas      fluxout      fluxin      fracmass
Carbon    3.7819E+11  3.7403E+11  4.5251E+13  4.5136E+13  1.0004E+00
          ifam       totddep     totwdep      totem
          3         4.346E+09   1.018E+10   1.080E+11
+++++
```



Miscellaneous 2: GenChem

- A note on GenChem
- Pre-processor
 - Reads ascii chemical equations and description files
 - Produced all CM_ files
- Needs code-clean and documentation before release. Will try to do in 2013...

• GenChem input



1.40e-12*exp(-1310*tinv)	□	NO + O ₃ = NO ₂ ;
KMT07		OH + NO = HONO ;
3.60e-12*exp(270*tinv)		H ₀₂ + NO = OH + NO ₂ ;
1.80e-11*exp(110*tinv)		NO + NO ₃ = 2.000 NO ₂ ;
1.40e-13*exp(-2470*tinv)		NO ₂ + O ₃ = NO ₃ ;
KMT08		OH + NO ₂ = HNO ₃ ;
KMT09		H ₀₂ + NO ₂ = H ₀₂ NO ₂ ;
KMT10		H ₀₂ NO ₂ = H ₀₂ + NO ₂ ;
3.20e-13*exp(690*tinv)		OH + H ₀₂ NO ₂ = NO ₂ ;
4.50e-14*exp(-1260*tinv)		NO ₂ + NO ₃ = NO + NO ₂ ;
KMT03		NO ₂ + NO ₃ = N ₂ O ₅ ;
KMT04		N ₂ O ₅ = NO ₂ + NO ₃ ;
2.00e-11		NO ₃ + OH = H ₀₂ + NO ₂ ;
.....		



Miscellaneous 3: Chemical schemes

- Have several mechanisms ready processed (almost):

Mechanism	Species	Reactions	Photochemical Reactions	Emitted VOCs (No. Biogenic)	Ref.
CRI v2	465	1202	185	116 (3)	Jenkin et al. (2008)
CRI v2 R5	195	569	96	3 (3)	Watson et al. (2008)
CBM-IV	38	95	13	10 (1)	Gery et al. (1989)
CB-05	70	189	27	16 (2)	Yarwood et al. (2005)
OSRM	70	197	25	15 (1)	Hayman et al. (2010)
EMEP-EmChem03 ^a	69	135		10 (1)	Simpson et al. (2003a), Andersson-Sköld and Simpson (1999)
EMEP-EmChem09 ^a	72	137	26	10 (1)	This work
EMEP-EmChem09soa	b	b	26	11 (2)	Bergström et al. (2012) ^c

- Please ask if interested.



Chemistry in EMEP model....

- Eliassen, A.; Hov, Ø., et al. A Lagrangian long-range transport model with atmospheric boundary layer chemistry J. Appl. Met., 1982, 21, 1645-1661
- Hov, Ø. Models of the chemical turnover in the atmospheric boundary layer, PhD, University of Oslo, 1981
- Isaksen, I. & Hov, Ø. Calculation of trends in the tropospheric concentration of O₃, OH, CO, CH₄ and NO_x Tellus, 1987, 39B, 271-285
- Simpson, D. Long period modelling of photochemical oxidants in Europe. Calculations for July 1985 Atmos. Environ., 1992, 26A, 1609-1634
- Simpson, D.; Andersson-Sköld, Y. & Jenkin, M. E. Updating the chemical scheme for the EMEP MSC-W oxidant model : current status The Norwegian Meteorological Institute, Oslo, Norway, 1993
- Simpson, D. Biogenic emissions in Europe 2: Implications for ozone control strategies J. Geophys. Res., 1995, 100, 22891-22906
- Kuhn, M.; et al., Intercomparison of the gas-phase chemistry in several chemistry and transport models Atm. Env., 1998, 32, 693-709
- Jonson, J.; et al., Chemical effects of UV fluctuations inferred from total ozone and tropospheric aerosol variations J. Geophys. Res., 2000, 105, 14561-14574B
- Andersson-Sköld, Y. & Simpson, D. Comparison of the chemical schemes of the EMEP MSC-W and the IVL photochemical trajectory models Atm. Env., 1999, 33, 1111-1129
- Simpson, D.; et al., The EMEP MSC-W chemical transport model -- technical description Atmos. Chem. Physics, 2012, 12, 7825-7865

And SOA:



- Andersson-Sköld, Y. & Simpson, D. Secondary organic aerosol formation in Northern Europe: a model study J. Geophys. Res., 2001, 106, 7357-7374
- Simpson, D.; Yttri, K.; Klimont, Z.; Kupiainen, K.; Caseiro, A.; Gelencsér, A.; Pio, C. & Legrand, M. Modeling Carbonaceous Aerosol over Europe. Analysis of the CARBOSOL and EMEP EC/OC campaigns J. Geophys. Res., 2007, 112, D23S14
- Simpson, D.; et al., Modelling SOA in EMEP: Experiments with the VBS Approach Transboundary Particulate Matter in Europe, Status Report 4/2004, The Norwegian Institute for Air Research (NILU), Kjeller, Norway, 2009
- Bergström, R.; et al., Modelling of organic aerosols over Europe (2002--2007) using a volatility basis set (VBS) framework: application of different assumptions regarding the formation of secondary organic aerosol Atmos. Chem. Physics, 2012, 12, 8499-8527
-

A wide-angle photograph of a desert landscape. In the foreground, a small group of people on horseback are riding across the sandy terrain. To the left, a large, rugged mountain peak rises sharply. The background features more desert land leading to a range of mountains with distinct horizontal sedimentary layers under a clear blue sky.

The end :-)