

EMEP MSC-W model

EMEP MSC-W model: Chemistry, status + changes

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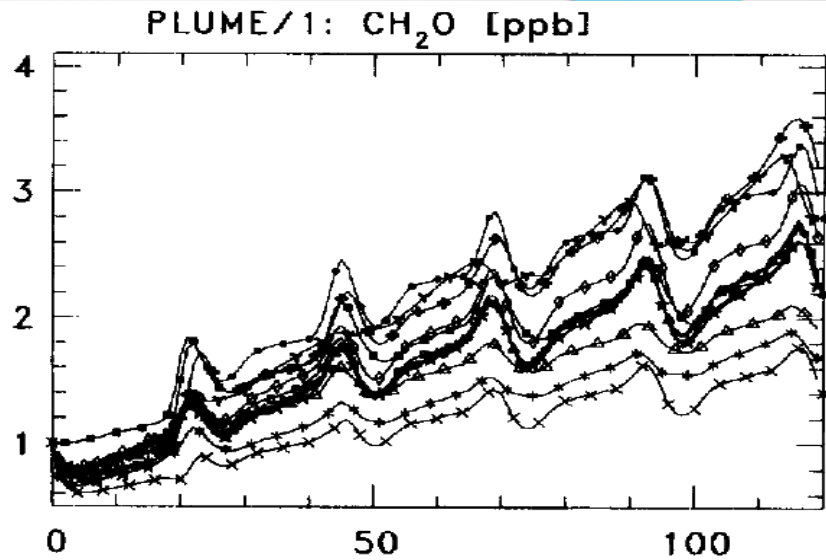
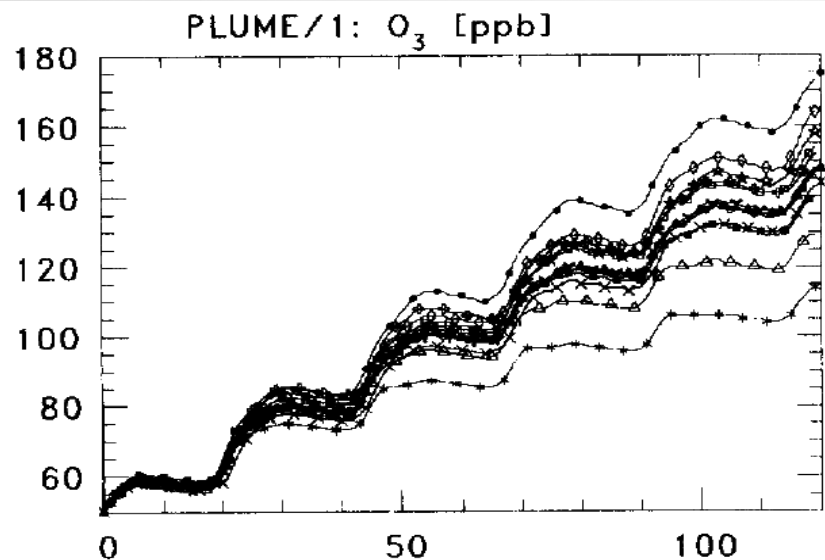
Outline:

- Gas-phase chemistry
- SOA module
- Code structure
- Recent updates
- Miscellaneous (GenChem, other schemes)

Gas-phase chemistry:

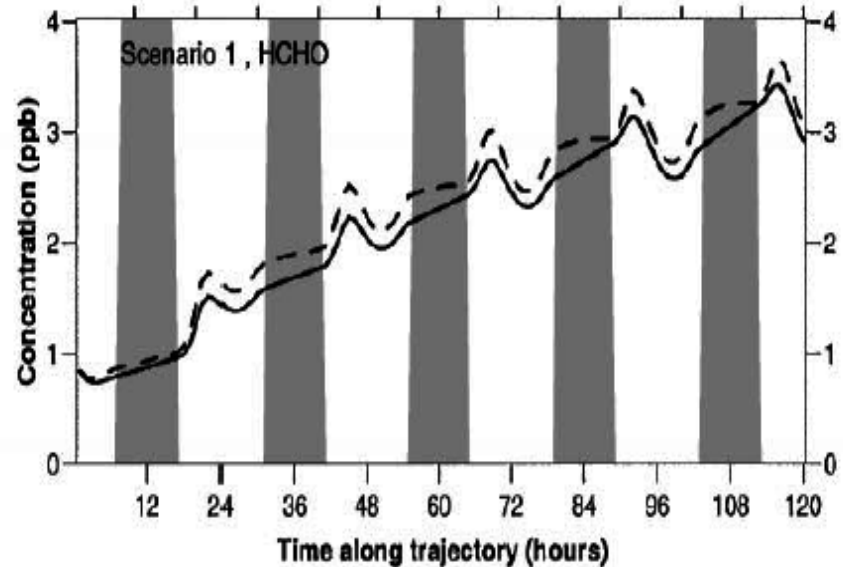
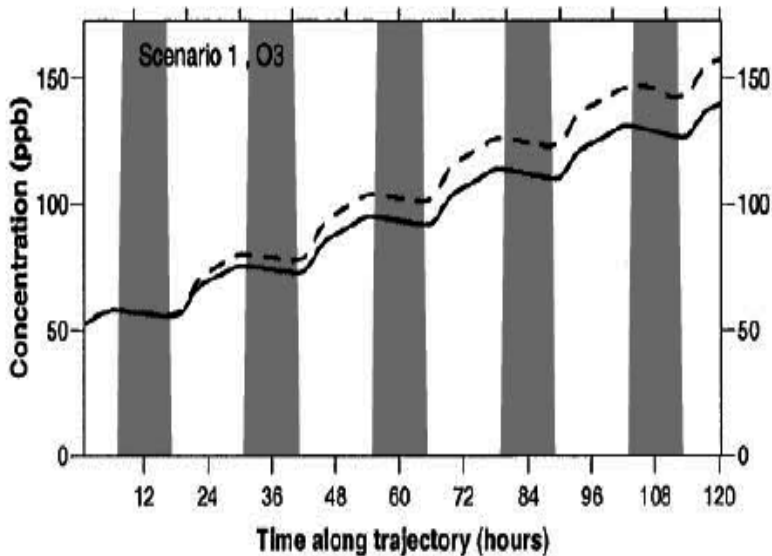
- EmChem09
 - 72 species
 - 137 reactions
 - 10 VOC (1 BVOC = isoprene)
- Update of previous EMEP chemical schemes, which date back to Eliassen, Hov, et al 1982
- 'Surrogate species' method, with relatively more RO₂ chemistry than e.g. CB schemes

Early comparisons.....



Kuhn et al., *Atm. Env.*, 1998, with EMEP, IVL, 4 x RADM, 3 x CB4, ...

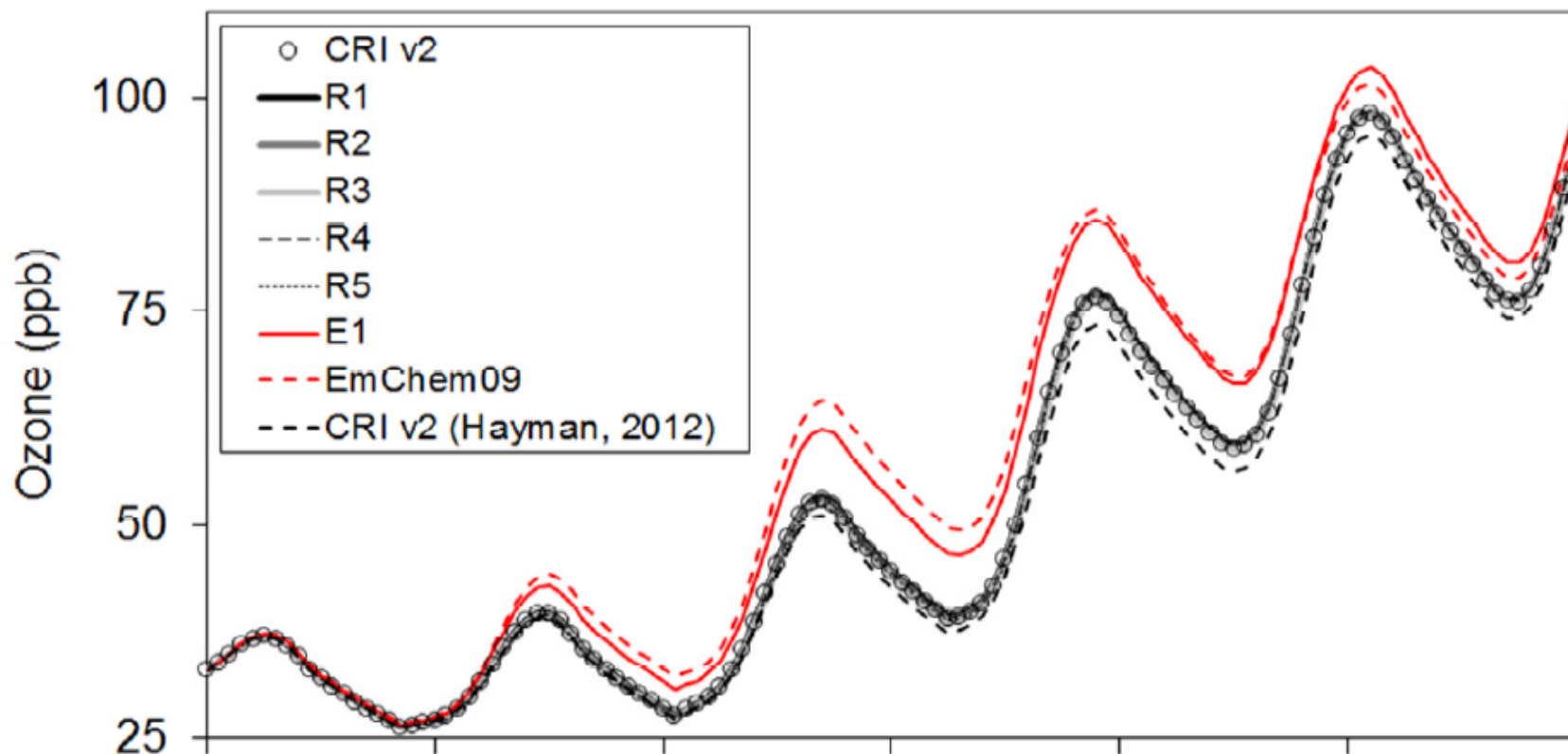
Follow-up: EMEP vs most complex (IVL): comparisons.....



Andersson-Sköld & Simpson, *Atm. Env.*, 1999, same setup as Kuhn et al. (+other cases)

- EMEP model compared well :-)

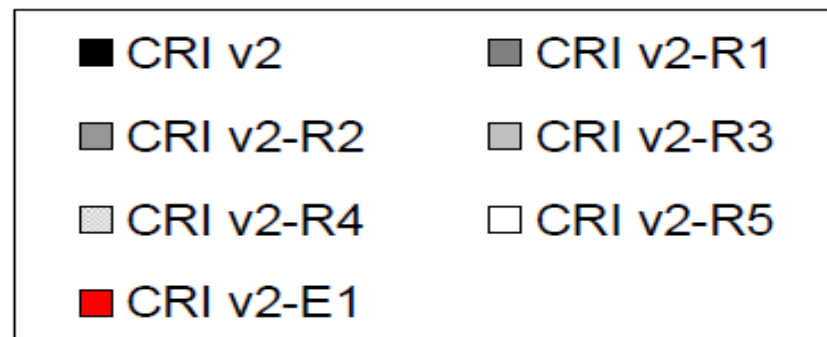
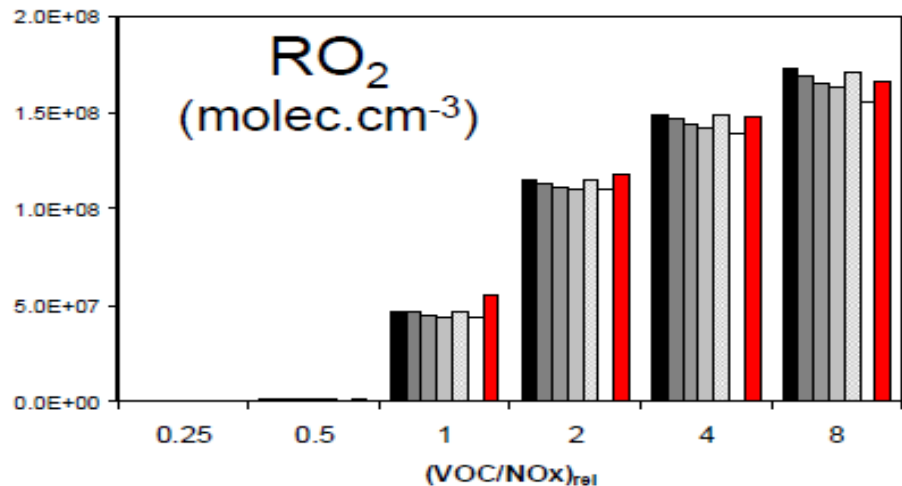
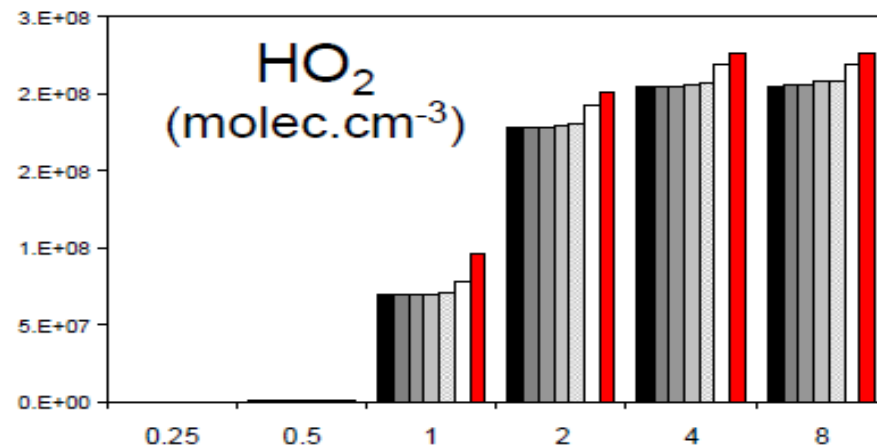
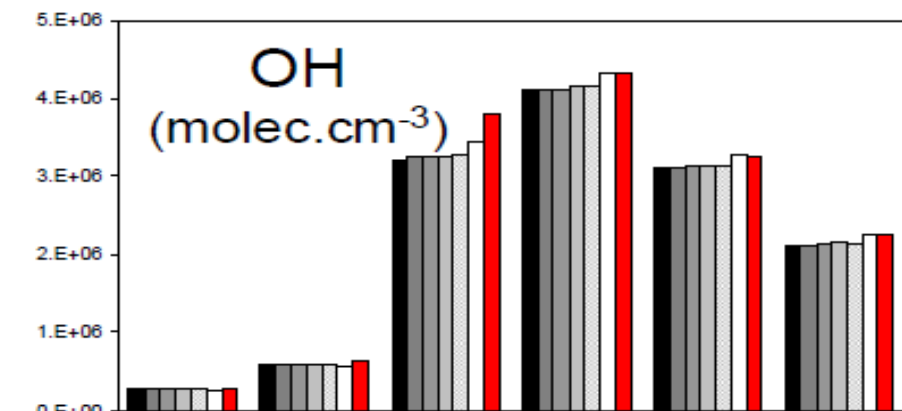
2009: Comparison with CRI model.



Ozone changes (c/o Mike Jenkin), simple plume

- CRI v2 ~ 1000 reactions, 112 AVOC

Comparison with CRI model.



Radicals (c/o Mike Jenkin)

- How do we handle chemistry in the model?

- CM_ files:

```
ChemFunctions_ml.f90
Chem_ml.f90
CM_BoundaryConditions.inc
CM_BoundaryConditions.inc.txt
CM_ChemGroups_ml.f90
CM_chempackages.txt
CM_ChemRates_ml.f90
CM_ChemSpecs_ml.f90
CM_DryDep.inc
CM_EmisBioNat.inc
CM_EmisFiles.inc
CM_emislist.csv
CM_EmisSpecs.inc
CM_Reactions1.inc
CM_Reactions2.inc
CM_WetDep.inc
Code_ml.f90
```


- CM_ChemSpecs_ml

- Species -
 - Number
 - Indices
 -

```
!>
module ChemSpecs_adv_ml
!-----
...
integer, public, parameter :: NSPEC_ADV = 126

integer, public, parameter ::
  IXADV_03      = 1 &
, IXADV_NO     = 2 &
!-----
end module ChemSpecs_adv_ml
!>

module ChemSpecs_shl_ml
!-----
...
  IXSHL_OD     = 1 &
, IXSHL_OP     = 2 &
!-----
...
module ChemSpecs_tot_ml
!-----
integer, public, parameter :: NSPEC_TOT = 142

integer, public, parameter ::
  OD           = 1 &
, OP          = 2 &
```

• CM_ChemSpecs_ml

• + characteristics

```
...  
contains  
subroutine define_chemicals()  
...  
!
```

			MW	NM	C	N	S	ExtC	C*	dH
species(O ₂)) = Chemical("O ₂ ")		16.0	0	0	0	0	0.0	0.0	0.0
species(O ₃)) = Chemical("O ₃ ")		16.0	0	0	0	0	0.0	0.0	0.0
species(OH)) = Chemical("OH ")		17.0	0	0	0	0	0.0	0.0	0.0
species(H ₂ O)) = Chemical("H ₂ O ")		33.0	0	0	0	0	0.0	0.0	0.0
species(O ₃)) = Chemical("O ₃ ")		48.0	0	0	0	0	0.0	0.0	0.0
species(NO)) = Chemical("NO ")		30.0	0	0	1	0	0.0	0.0	0.0
species(NO ₂)) = Chemical("NO ₂ ")		46.0	0	0	1	0	0.0	0.0	0.0
species(PAN)) = Chemical("PAN ")		121.0	0	2	1	0	0.0	0.0	0.0
species(NO ₃)) = Chemical("NO ₃ ")		62.0	0	0	1	0	0.0	0.0	0.0

•CM_Reactions1.inc

! -> 03

```
P = &  
  rct(1,k) * xnew(OP ) &  
  + 0.15*rct(37,k) * xnew(CH3C002 ) * xnew(H02 ) &  
  + 0.29*rct(60,k) * xnew(MAC03 ) * xnew(H02 )
```

```
L = &  
  rct(5,k)* xnew(N0 ) &  
  + rct(6,k)* xnew(N02 ) &  
  + rct(7,k)* xnew(OH ) &  
  + rct(8,k)* xnew(H02 ) &  
  + AQRCK(ICLRC2,K)* xnew(S02 ) &  
  + rct(43,k)* xnew(C2H4 ) &  
  + rct(44,k)* xnew(C3H6 ) &  
  + rct(50,k)* xnew(C5H8 ) &  
  + rct(53,k)* xnew(MACR ) &  
  + rct(58,k)* xnew(MVK ) &  
  + rcphot(IDA03,K) &  
  + rcphot(IDB03,K)
```

```
xnew(03)= ( xold(03) + dt2 * P) /(1.0 + dt2*L )
```

• CM_ChemGroups_ml

```
!> _____ <

- module ChemGroups_ml
!-----

use ChemSpecs_tot_ml ! => species indices
use OwnDataTypes_ml ! => typ_sp
implicit none
private
! Assignment of groups from GenIn.species:
public :: Init_ChemGroups

! ----- Gas/particle species -----
..

integer, public, parameter :: INDEX_WDEP_OXN_GROUP = 2
integer, public, target, save, dimension(4) :: &
    WDEP_OXN_GROUP = (/ HN03,HONO,N03_F,N03_C /)

integer, public, parameter :: INDEX_WDEP_PPM10_GROUP = 3
integer, public, target, save, dimension(11) :: &
    WDEP_PPM10_GROUP = (/ POM_F_WOOD,POM_F_FFUEL,
UEL_NEW,EC_F_FFUEL_AGE,EC_C_FFUEL,REMPPM25,REMPPM_C /)
```

OA, POA, SOA ;-)

OA: Subject=Horrendous!

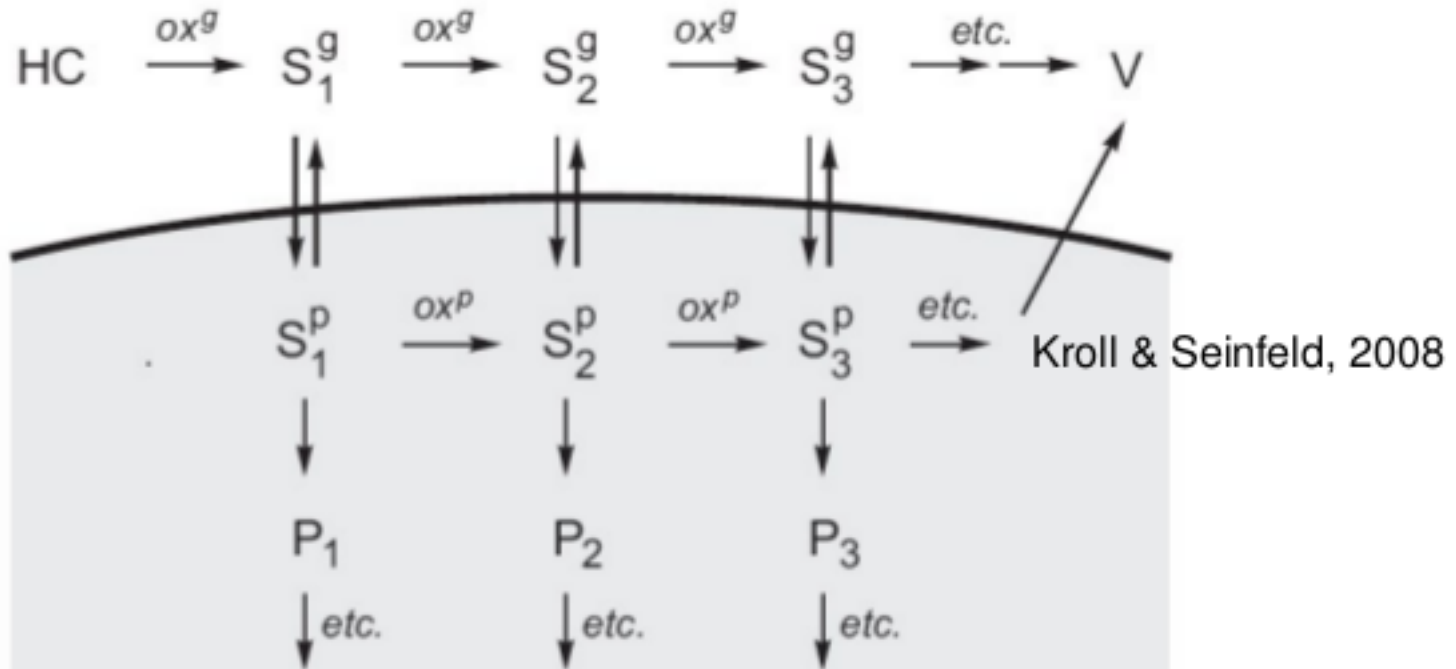
Some issues:

- Which precursors? (Isoprene, terpenes, aromatics ..)
- 1000s of compounds, mainly unknown.
- Gas-phase? Aqueous-phase? Particle-phase?
- Thermodynamics/Volatility
- Aging
- Artifact-rich observations!

See e.g. Donahue et al., Atmos. Environ., 2009, Hallquist et al., ACP, 2009

OA cont,

- Need to deal with equilibria, e.g.



OA cont., “VBS”:

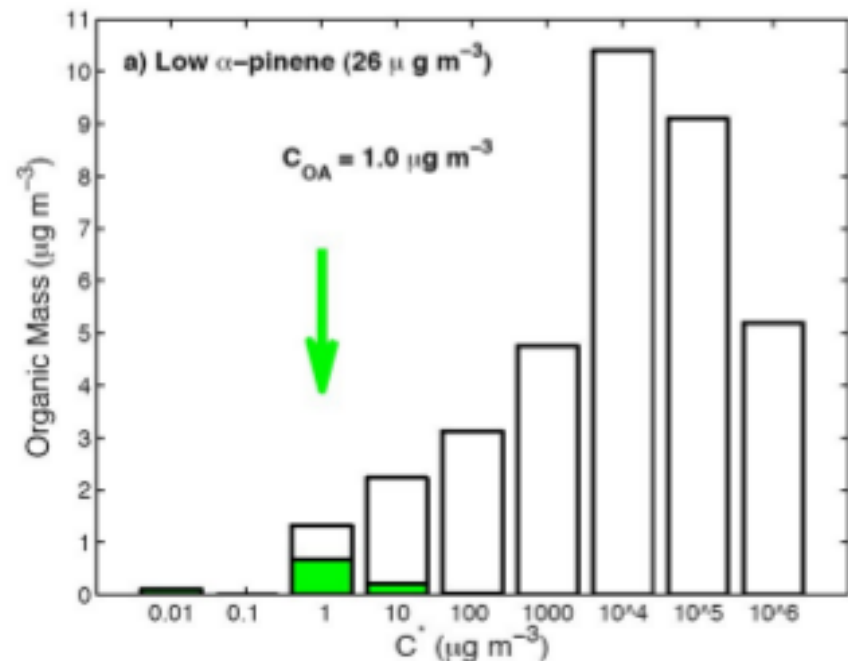
- Volatility Basis Set (Donahue and colleagues)

- Gas-Particle partitioning:

$$\frac{A_i}{G_i} = \frac{C_{OA}}{C_i^*}$$

where

- C_i^* is saturation concentration,
= $f(\text{Vapour pressure})$



OA in model code:

- Code tracks species, not phase, for advection
- Gas/aerosol fractions calculated using partitioning theory in My_SOA_ml
- Affects deposition, aging reactions etc.
-

OA in model code:

- Origin (GenChem) of reactions looks like:

```
1.2e-11*exp(444.0*TINV)    [OH] + apinene = TERPPeroxy
[
KR02NO TERPPeroxy + NO    = N02 + 0.08 BSOC_ug1 + 0.672 NON_C_BSOA_ug1
                             + 0.813333 BSOC_ug10 + 6.832 NON_C_BSOA_ug10
                             + 1.34 BSOC_ug1e2 + 11.256 NON_C_BSOA_ug1e2
                             + 3.333333 BSOC_ug1e3 + 28. NON_C_BSOA_ug1e3
~
```

- - ie produces range of products of different volatility

OA in CM_Reactions2:

- Gas-phase chemistry

```
! -> BSOC_UG1
```

```
P = &  
    0.002833*rct(40,k)*xnew(ISR02)*xnew(NO) &  
    + 0.0255*rct(47,k)*xnew(ISR02)*xnew(HO2) &  
    + 0.08*rct(40,k) * xnew(TERPPEROXY ) * xnew(NO ) &  
    + 0.715333*rct(75,k) * xnew(TERPPEROXY ) * xnew(HO2 ) &  
    + rct(85,k)*xnew(OH) * xnew(BSOC_UG10 )
```

```
L = &  
    rct(84,k)*xnew(OH)
```

```
xnew(BSOC_UG1)= ( xold(BSOC_UG1) + dt2 * P ) / ( 1.0 + dt2*L )
```

OA final words:

- OA schemes can give many very different answers.

The basic rule:



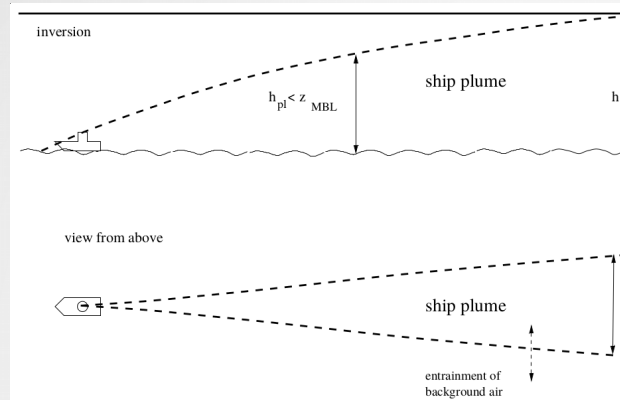
Garbage in ⇒ ⇒ Garbage out:

SOA twist:



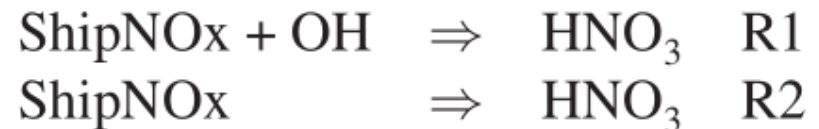
Garbage in the middle!

2015 Changes 1: ShipNOx

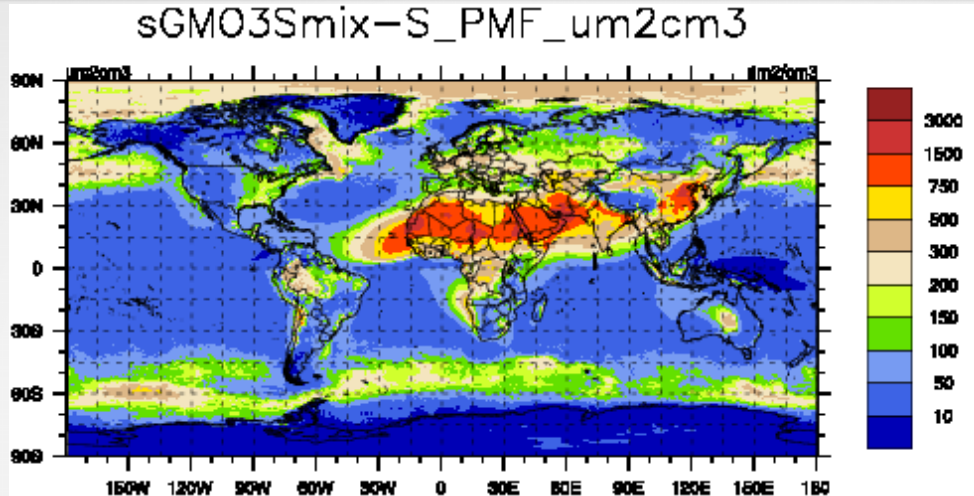


- ShipNO_x:
- Pragmatic solution to provide faster conversion to HNO₃ and reduce O₃ production over oceans.
- Loosely (very!) based on GEOS-Chem PARANOX (Vinken et al, 2011)
- 50% of ship NO emitted as 'ShipNO_x'

- Non-linearity of ozone chemistry can cause ship plumes to
 - Generate too much O₃
 - Give too long lifetime for NO_x



2015 Changes 2: Aerosol surface area



- Aerosol surface area (S) now calculated using 'Gerber' functions
 - Empirical, we like ;-)
 - Avoids too much reliance on e.g. MARS/ISORROPIA
 - Unclear if high S due to fine-dust, sea-salt is correct
 - Large implications for e.g. modelling in Asia!

2015 Changes 3: gas-aerosol interactions

- Gas-aerosol rates now depend on surface area
- Replaces older & cruder system
- Links inorganic chemistry to dust & sea-salt

$$k_x = \frac{1}{4} c_x S \gamma_x$$

Table 1.1: Main aerosol-uptake reactions in rv4.7 EMEP model

Reaction	γ
$\text{N}_2\text{O}_5 \text{ (g)} \longrightarrow 2 \text{HNO}_3 \text{ (aq)}$	Options: Smix, SmixTen, or fixed values. See Sect.1.2.1
$\text{HNO}_3 \text{ (g)} + \text{sea-salt}_c \longrightarrow \text{NO}_{3,c}$	0.01
$\text{HNO}_3 \text{ (g)} + \text{dust}_c \longrightarrow \text{NO}_{3,c}$	0.02
$\text{HO}_2 \text{ (g)} + \text{PM}_c \longrightarrow \frac{1}{2} \text{H}_2\text{O}$	0.2
$\text{O}_3 \text{ (g)} + \text{dust}_c \longrightarrow \text{HO}_2$	1.0×10^{-6}

Notes: $\text{NO}_{3,c}$ represents coarse mode nitrate. These uptake reactions are applied whenever RH exceeds 40 %.

Options for N2O5 hydrolysis

- $\text{N}_2\text{O}_5 + \text{H}_2\text{O} \rightarrow 2 \text{HNO}_3$
- Important loss of N_x in troposphere!
- In config_emep.nml
 - USES
%n2O5hydrolysis =
'SmixTen'

- Options:

- Smix

$$\gamma = f_{\text{SIA}} \gamma_{\text{SIA}} + f_{\text{OM}} \gamma_{\text{OM}} + f_{\text{SS}} \gamma_{\text{SS}} + f_{\text{du}} \gamma_{\text{du}}$$

- **SmixTen = Smix/10**

- Gamma:0.002

- These rates are **very** uncertain and will likely be revised in future. Feel free to test

Miscellaneous 1: Mass balance

- Please check : fracmass in RunLog

```
+++++
Mass balance 1 Sulphur
]+++++
          sumint      summas      fluxout      fluxin      fracmass
Sulphur  3.4648E+07  4.2275E+07  5.2910E+09  3.6425E+09  1.0117E+00
  ifam    totddep      totwdep      totem
    1      1.795E+09    4.573E+09    7.888E+09
+++++
Mass balance 2 Nitrogen
+++++
          sumint      summas      fluxout      fluxin      fracmass
Nitrogen 7.3405E+07  1.3389E+08  1.1912E+10  8.8174E+09  1.0078E+00
  ifam    totddep      totwdep      totem
    2      5.729E+09    1.245E+10    2.110E+10
+++++
Mass balance 3 Carbon
+++++
          sumint      summas      fluxout      fluxin      fracmass
Carbon   3.7819E+11  3.7403E+11  4.5251E+13  4.5136E+13  1.0004E+00
  ifam    totddep      totwdep      totem
    3      4.346E+09    1.018E+10    1.080E+11
+++++
```


Miscellaneous 2: GenChem

- A note on GenChem
- Pre-processor
 - Reads ascii chemical equations and description files
 - Produced all CM_ files
- 2013 statement: Needs code-clean and documentation before release. Will try to do in 2013...
- 2015 status: Code was 'cleaned' – converted to python for ESX model. Needs to be re-implemented in EMEP and documented (still!)

• GenChem input

```
1.40e-12*exp(-1310*tinv)  □ NO + O3 = N2O ;
KMT07                    OH + NO = HONO ;
3.60e-12*exp(270*tinv)    H2O + NO = OH + N2O ;
1.80e-11*exp(110*tinv)    NO + N2O = 2.000 N2O ;
1.40e-13*exp(-2470*tinv)  N2O + O3 = N2O ;
KMT08                    OH + N2O = HNO3 ;
KMT09                    H2O + N2O = H2ON2O ;
KMT10                    H2ON2O = H2O + N2O ;
3.20e-13*exp(690*tinv)    OH + H2ON2O = N2O ;
4.50e-14*exp(-1260*tinv)  N2O + N2O = NO + N2O ;
KMT03                    N2O + N2O = N2O5 ;
KMT04                    N2O5 = N2O + N2O ;
2.00e-11                  N2O + OH = H2O + N2O ;
.....
```

Miscellaneous 3: Chemical schemes

- Have several mechanisms ready processed (almost):

Mechanism	Species	Reactions	Photochemical Reactions	Emitted VOCs (No. Biogenic)	Ref.
CRI v2	465	1202	185	116 (3)	Jenkin et al. (2008)
CRI v2 R5	195	569	96	3 (3)	Watson et al. (2008)
CBM-IV	38	95	13	10 (1)	Gery et al. (1989)
CB-05	70	189	27	16 (2)	Yarwood et al. (2005)
OSRM	70	197	25	15 (1)	Hayman et al. (2010)
EMEP-EmChem03 ^a	69	135		10 (1)	Simpson et al. (2003a), Andersson-Sköld and Simpson (1999)
EMEP-EmChem09 ^a	72	137	26	10 (1)	This work
EMEP-EmChem09soa	b	b	26	11 (2)	Bergström et al. (2012) ^c

Miscellaneous 4: odds n' ends

- The model has some 'untried' components
- ISORROPIA instead of MARS/EQSAM
- PointSource_ml
 - Allows stacks, plume rise, gaussian spread (in vertical)
 - (cooperation with NILU/M. Karl)
 - 'probably works' but not 100% integrated in mass budget
- Use at own risk! Please ask!

Papers: chemistry in EMEP model....

- **Eliassen, A.; Hov, Ø., et al. A Lagrangian long-range transport model with atmospheric boundary layer chemistry J. Appl. Met., 1982, 21, 1645-1661**
- Hov, Ø. Models of the chemical turnover in the atmospheric boundary layer, PhD, University of Oslo, 1981
- Isaksen, I. & Hov, Ø. Calculation of trends in the tropospheric concentration of O₃, OH, CO, CH₄ and NO_x Tellus, 1987, 39B, 271-285
- Simpson, D. Long period modelling of photochemical oxidants in Europe. Calculations for July 1985 Atmos. Environ., 1992, 26A, 1609-1634
- Simpson, D.; Andersson-Sköld, Y. & Jenkin, M. E. Updating the chemical scheme for the EMEP MSC-W oxidant model : current status The Norwegian Meteorological Institute, Oslo, Norway, 1993
- Simpson, D. Biogenic emissions in Europe 2: Implications for ozone control strategies J. Geophys. Res., 1995, 100, 22891-22906
- Kuhn, M.; et al., Intercomparison of the gas-phase chemistry in several chemistry and transport models Atm. Env., 1998, 32, 693-709
- Jonson, J.; et al., Chemical effects of UV fluctuations inferred from total ozone and tropospheric aerosol variations J. Geophys. Res., 2000, 105, 14561-14574B
- Andersson-Sköld, Y. & Simpson, D. Comparison of the chemical schemes of the EMEP MSC-W and the IVL photochemical trajectory models Atm. Env., 1999, 33, 1111-1129
- Simpson, D.; et al., The EMEP MSC-W chemical transport model -- technical description Atmos. Chem. Physics, 2012, 12, 7825-7865

And SOA:

- Andersson-Sköld, Y. & Simpson, D. Secondary organic aerosol formation in Northern Europe: a model study J. Geophys. Res., 2001, 106, 7357-7374
- Simpson, D.; Yttri, K.; Klimont, Z.; Kupiainen, K.; Caseiro, A.; Gelencsér, A.; Pio, C. & Legrand, M. Modeling Carbonaceous Aerosol over Europe. Analysis of the CARBOSOL and EMEP EC/OC campaigns J. Geophys. Res., 2007, 112, D23S14
- Simpson, D.; et al., Modelling SOA in EMEP: Experiments with the VBS Approach Transboundary Particulate Matter in Europe, Status Report 4/2004, The Norwegian Institute for Air Research (NILU), Kjeller, Norway, 2009
- Bergström, R.; et al., Modelling of organic aerosols over Europe (2002--2007) using a volatility basis set (VBS) framework: application of different assumptions regarding the formation of secondary organic aerosol Atmos. Chem. Physics, 2012, 12, 8499-8527
- Bergström, R.; Hallquist, M.; Simpson, D.; Wildt, J. & Mentel, T. F. Biotic stress: a significant contributor to organic aerosol in Europe? Atmospheric Chemistry and Physics, 2014, 14, 13643-13660

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We 'bugged' the EMEP model ;-)

- Bergström et al, ACP, 2014.
- - unplanned' cooperation with Thomas Mentel, arising from a breakfast chat in earlier ECLAIRE meeting.
- - mixed models, smog-chambers and
- statistics on honey production

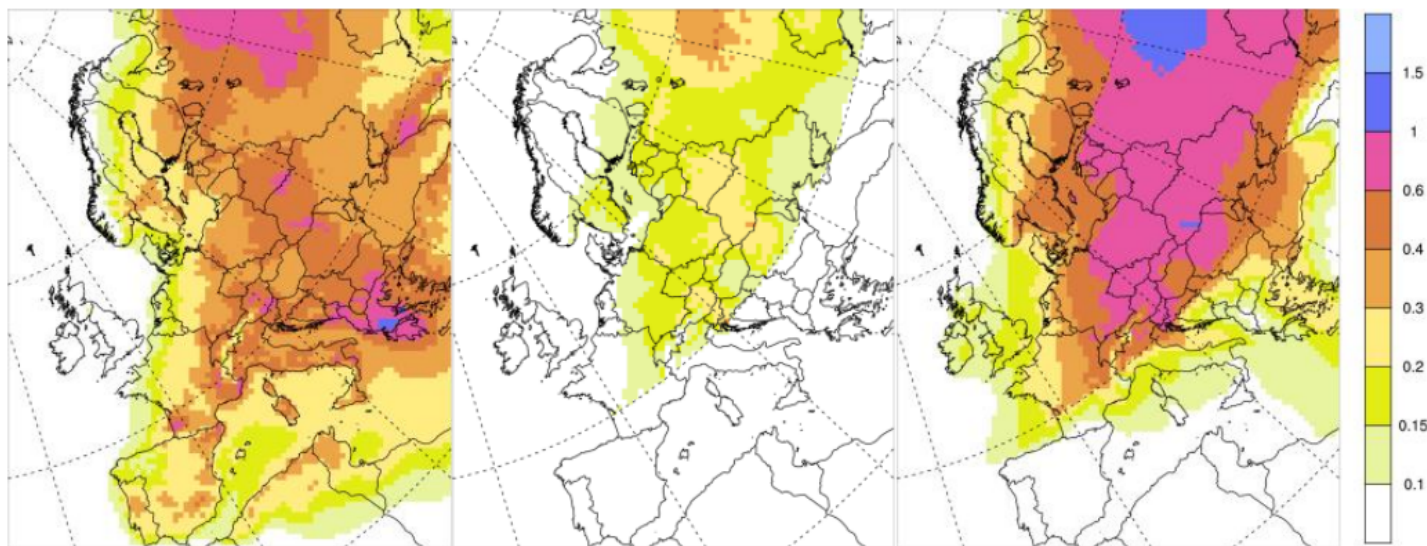
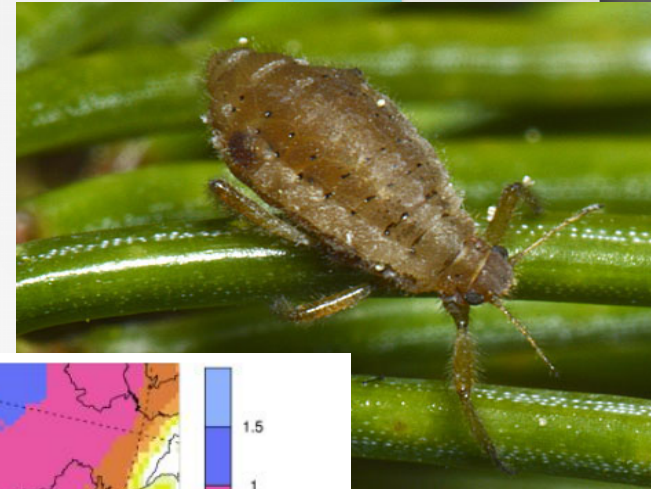


Figure SIE1. Model calculated 6-month mean (Apr-Sep) concentrations of biogenic SOA and biotic stress-induced OA (SIE-OA); **(a)** BSOA from constitutive emissions (reference simulation, Case 0), **(b)** SIE-OA in Case 1 (biotic stress with sesquiterpene (SQT) emissions), **(c)** SIE-OA in Case 2 (biotic stress with emissions of SQT and methyl salicylate). Unit: $\mu\text{g m}^{-3}$.

Almost final...

- The chemical schemes are getting more and more complex, e.g. SOA, aerosol schemes
- Difficult to summarise or address all issues related to EMEP model
- If interested in e.g. OA, or GenChem:
 - Please ask !



The end :-)

